Barnsley Academy – Y11- Combined Science C7- Scheme of work Week 1 – 2023-24

Term – Week						
	1	2	3	4		
Lesson Focus	Crude oil and Alkanes	Fractional distillation	Cracking	Using the products of crude oil		
Prerequisite Knowledge	Covalent bonding Properties of covalent compounds	 Boiling points (C2) Distillation (C1) 	Alkanes (L1-C7)	- Complete and incomplete combustion		
Core Knowledge	 Describe the composition of crude oil Draw the first five alkanes and write their chemical formula Use the general chemical formula of alkanes Describe and explain how alkane chain length affects its boiling point 	 Describe how crude oil is separated into fractions Describe trends in the physical and chemical properties of the fractions collected 	 Describe the process and products of cracking Describe the test for alkenes and its positive result Explain why cracking is necessary Represent cracking using equations 	 Draw and name polymers made from alkenes Compare complete and incomplete combustion Calculate reacting masses for combustion 		
Expert Model /Guided Practice/Agreed Approach (Procedural Knowledge)	 How to calculate chemical formula of alkanes 	- Steps for fractional distillation	Steps for cracking equations	 Steps for drawing and naming polymers Calculating reacting masses steps 		
Independent Practice	IP1- Describe composition of alkanes IP2- Alkanes exam q IP3- Boiling point graphs exam q	IP1- Fractional distillation exam q	IP1- Describe what cracking is and why it is useful IP2- Cracking exam q	IP1- Polymer exam q IP2- Comparing complete and incomplete combustion IP3- Calculating reacting masses		

Assessment (Informal/Formal)	 Whiteboard checks Circulation Live marking Exam q 	 Whiteboard checks Circulation Live marking Exam q 	 Whiteboard checks Circulation Live marking Exam q 	 Whiteboard checks Circulation Live marking Exam q
Resources			https://www.youtube.com/watch?v=Ehcq4cARkMI	
Specific SEN(D)/EAL support				